

CHEM*3570 Analytical Biochemistry Centrifugation

1. For a cylinder, the moment of inertia $I = \frac{1}{2} mr^2$ (see any Physics textbook), and kinetic energy of rotation $E = \frac{1}{2} I\omega^2$. Consider an ultracentrifuge rotor to be a cylinder of titanium (density = 4.5), diameter = 25 cm., height = 10 cm. Calculate the kinetic energy of such a rotor, spinning at 50,000 rpm. How fast would you have to drive a 1,000 kg truck, to acquire the equivalent kinetic energy?

$$I = \frac{1}{2} mr^2 \quad r = 12.5 \text{ cm} \quad h = 10 \text{ cm.} \quad \therefore V = \pi r^2 h = 4,909 \text{ cm}^3$$

$$\rho = 4.5 \text{ gm cm}^{-3} \quad \therefore m = 22 \text{ kg} \quad I = \frac{1}{2} (22 \text{ kg}) (1.25 \times 10^{-1} \text{ m})^2 = 0.172 \text{ kg m}^2$$

$$E = \frac{1}{2} I \omega^2 \quad 50,000 \text{ rpm} = 833 \text{ rps} \quad \therefore \omega = 2\pi 833 \text{ rps} = 5,236 \text{ s}^{-1}$$

(Note that ω is given in radians per second)

$$E = \frac{1}{2} (0.172 \text{ kg m}^2) (27.4 \times 10^6 \text{ s}^{-2}) = 2.36 \times 10^6 \text{ kg m}^2 \text{ s}^{-2} = 2.36 \text{ MJ}$$

$$1 \text{ ton} = 10^3 \text{ kg} \quad E = \frac{1}{2} mv^2. \quad \therefore 2.36 \times 10^6 \text{ kg m}^2 \text{ s}^{-2} = 500 \text{ kg v}^2$$

$$\therefore v^2 = (2.36 \times 10^6) / (500) = 4.6 \times 10^3 \text{ m}^2 \text{ s}^{-2} \quad \therefore v = 68 \text{ m s}^{-1}$$

That's about 250 kph, so you can see why ultracentrifuges are armour-plated!

2. A protein is studied in the analytical ultracentrifuge. At high speed (60,000 rpm), the sedimentation velocity v is measured as $1.58 \times 10^{-4} \text{ cm sec}^{-1}$ at a radius of 10.0 cm.

Estimate the sedimentation coefficient of the protein, expressing the results in Svedberg Units (S) where $1 \text{ S} = 10^{-13} \text{ sec}$. If the sample cell in the ultracentrifuge is 1.5 cm deep, estimate approximately how long it takes to sediment from top to bottom.

$$s = \frac{v}{\omega^2 r} = \frac{1.58 \times 10^{-4} \text{ cm s}^{-1}}{\left(\frac{2\pi \times 60,000}{60}\right)^2 \text{ s}^{-2} \times 10.0 \text{ cm}} = 4.0 \times 10^{-13} \text{ s}$$

$$\therefore s = 4.00 \text{ S}$$

Time taken to sediment 1.5 cm: First, an *approximate* calculation, assuming constant velocity:

$$t = \frac{1.5 \text{ cm}}{1.58 \times 10^{-4} \text{ cm s}^{-1}} = 9,500 \text{ s} = 2 \text{ h } 38 \text{ m}$$

Actually, as the molecule sediments, it becomes further from the center of rotation; the g force increases with radius. Therefore, the time to sediment is more accurately given by solving the differential equation:

$$v = \frac{dr}{dt} = s\omega^2 r \quad \int_{r_1}^{r_2} \frac{dr}{r} = s\omega^2 \int_0^t dt$$

r_1 and r_2 are the initial radius and radius at time t , respectively.

$$\ln \frac{r_2}{r_1} = s\omega^2 t \quad t = \frac{\ln\left(\frac{r_2}{r_1}\right)}{s\omega^2}$$

The factor $k = [\ln(r_2/r_1)]/\omega^2$ is an indicator of the required sedimentation time for a given rotor:

$$t = k/s$$

For: $r_1 = 10$ cm and $r_2 = 11.5$ cm: $\ln(r_2/r_1) = 0.14$.

$$t = 0.14 / s\omega^2 = 0.14 / \{(4 \times 10^{-13} \text{ s}) (2\pi \times 10^3 \text{ s}^{-1})^2\} = 8,860 \text{ sec.}$$

3. The frictional coefficient f for a molecule is not easy to predict, because it depends on the detailed shape and surface properties of the molecule. However, for a macroscopic hard sphere, we can apply Stokes' Law (1851): $f = 6\pi\eta r$

where η = viscosity of the solution and r is the radius of the particle.

A plastic ball of radius 1 cm. is dropped through a graduated cylinder full of cooking oil, $\eta = 20$ Pa-s (N s m^{-2}). The density of the ball is 0.95 and the density of the oil is 0.8. At what velocity will the ball descend?

Note: Units of η : $\text{N s m}^{-2} = \text{kg m s}^{-2} \text{ s m}^{-2} = \text{kg s}^{-1} \text{ m}^{-1}$

$$f = 6\pi\eta r \quad \text{drag force} = fv \quad \text{mass} = \rho V = \rho(4/3\pi r^3) \quad \text{weight} = mg$$

Let σ = density of liquid. buoyancy = σV ; effective mass = $(\rho - \sigma)V$

At limiting velocity: $fv = m_{\text{eff}}g \therefore v = m_{\text{eff}}g/f$

$$v = \frac{m_{\text{eff.}} g}{f} = \frac{\frac{4}{3} \pi r^3 (\rho - \sigma) g}{6 \pi \eta r} = \frac{2r^2 (\rho - \sigma) g}{9 \eta}$$

$$v = \frac{2r^2 (\rho - \sigma) g}{9 \eta} = \frac{2(10^{-2} \text{ m})^2 (.95 - .8) \times 10^3 \text{ kg m}^{-3} (9.8 \text{ ms}^{-2})}{9(20 \text{ kg s}^{-1} \text{ m}^{-1})}$$

$$= 1.6 \times 10^{-3} \text{ m s}^{-1} = 1.6 \text{ mm s}^{-1}$$

4. As a simple model, we may assume that proteins are simply hard spheres of uniform density. In this case, how will the sedimentation coefficient s vary as a function of m_0 ?

$$s = \frac{m_0 (1 - \bar{v} \rho)}{f}$$

$$m_0 \propto r^3 \therefore f \propto r \propto (m_0)^{1/3}$$

$$s \propto \frac{m_0}{f} \propto \frac{m_0}{(m_0)^{1/3}} \propto (m_0)^{2/3}$$

5. The Meselson-Stahl experiment which proved the semi-conservative model of DNA replication relied on the analysis of *E. coli* DNA labelled with the stable isotope ^{15}N (by growing the bacteria on medium made with $^{15}\text{NH}_4\text{Cl}$). What is the expected difference in density between ^{15}N and ^{14}N DNA?

Number of nitrogen atoms in bases: A: 5; C: 3; G: 5; T: 2; average = 7.5 N atoms per base pair.

Average MW of base pair \approx 660 Da.

So $\Delta_{\text{mass}} = 7.5 \text{ units} / 660 \approx 1\%$ difference between fully ^{15}N -labelled and natural DNA; half this difference for the semi-conservative product (one strand ^{15}N and one strand ^{14}N).